M1. $\quad \mathbf{X}$ is $\mathrm{CH}_{3} \mathrm{CN}$ or ethanenitrile or ethanonitrile or methyl cyanide or cyanomethane or ethyl nitrile or methanecarbonitrile

Not ethanitrile but contradiciton of name and structure lose marks
$\mathbf{Y}$ is $\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{NH}_{2}$ or ethylamine or aminoethane or ethanamine

Step 1: reagent KCN not $\mathrm{HCN} / \mathrm{HCl}$ condition (aq)/alcohol - only allow condition if reagent correct or incomplete

$\mathbf{Z}$ is an amine or aminoalkane or named amine even if incorrect name for $\mathbf{Z}$ secondary (only award if amine correct)

$\left(\mathrm{Br}^{-}\right)+$can be on $N$ or outside brackets as shown
nucleophilic substitution

M2. (a)


Further reaction / substitution / formation of $2^{\circ} / 3^{\circ}$ amines etc (1) use an excess of $\mathrm{NH}_{3}$ (1)
(b) repels nucleophiles (such as $\mathrm{NH}_{3}$ )
(1)


## Notes

(a) allow $\mathrm{S}_{\mathrm{N}} 1$
penalise: $\mathrm{Br}^{-}$intead of $\mathrm{NH}_{3}$ removing $\mathrm{H}^{+}$for M 4
not contamination with other amines (this is in the question) not diamines
(b) allow because $\mathrm{NH}_{3}$ is a nuclephile or benzene is (only) attacked by electrophiles or $\mathrm{C}-\mathrm{Br}$ bond (in bromobenzene) is stronger / less polar or Br Ip delocalized
$\mathrm{HNO}_{3} / \mathrm{H}_{2} \mathrm{SO}_{4}$ without either conc scores (1) allow $20-60^{\circ}$ for (1) (any 2 ex 3 )
allow name or structure of nitrobenzene
other reducing agents: Fe or Sn with HCl (conc or dil or neither)
not conc $\mathrm{H}_{2} \mathrm{SO}_{4}$ or conc $\mathrm{HNO}_{3}$
allow $\mathrm{Ni} / \mathrm{H}_{2}$
Not $\mathrm{NaBH}_{4}$ or $\mathrm{LiAlH}_{4}$
ignore wrong descriptions for reduction step e.g. hydrolysis or hydration

Organic points
(1) Curly arrows: must show movement of a pair of electrons, i.e. from bond to atom or from Ip to atom / space
e.g.


OR

(2) Structures
penalise sticks (i.e.


 or $\quad-\mathrm{NH}_{2}$
or






Penalise once per paper


